

Vitamin K

Vitamin K (K from "Koagulations-Vitamin" in German and Scandinavian languages^[1]) denotes a group of lipophilic, hydrophobic vitamins that are needed for the posttranslational modification of certain proteins, mostly required for blood coagulation, but also a number of other proteins that chelate calcium ions and are involved in bone and other tissue metabolism. They are 2-methyl-1,4-naphthoquinone derivatives.

Vitamin K₁ is also known as phylloquinone or phytonadione (also called phytonadione). Vitamin K₂ (menaquinone, menatetrenone) is normally produced by bacteria in the large intestine,^[2] and dietary deficiency is extremely rare unless the intestines are heavily damaged, are unable to absorb the molecule, or are subject to decreased production by normal flora, as seen in broad spectrum antibiotic use.^[3]

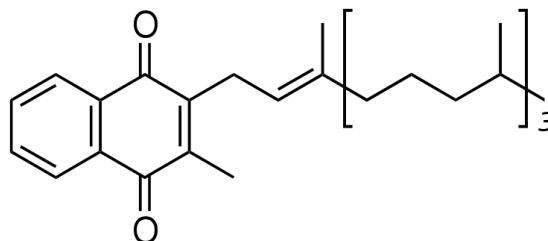
There are three synthetic forms of vitamin K, vitamins K₃, K₄, and K₅, which are used in many areas including the pet food industry (vitamin K₃) and to inhibit fungal growth (vitamin K₅).^[4]

Chemical structure

All members of the vitamin K group of vitamins share a methylated naphthoquinone ring structure, and vary in the aliphatic side chain attached at the 3-position (see figure 1). Phylloquinone (also known as vitamin K₁) invariably contains in its side chain four isoprenoid residues, one of which is unsaturated.

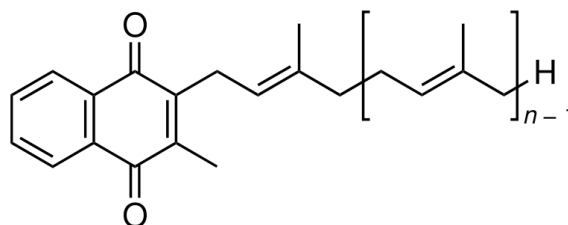
Menaquinones have side chains composed of a variable number of unsaturated isoprenoid residues; generally they are designated as MK-n, where n specifies the number of isoprenoids.

It is generally accepted that the naphthoquinone is the functional group, so that the mechanism of action is similar for all K-vitamins. Substantial differences may be expected, however, with respect to intestinal absorption, transport, tissue distribution, and bio-availability. These differences are caused by the different lipophilicity of the various side chains, and by the different food matrices in which they occur.



Vitamin K₁ (phylloquinone). Both forms of the vitamin contain a functional naphthoquinone ring and an aliphatic side chain.

Phylloquinone has a phytyl side chain.



Vitamin K₂ (menaquinone). In menaquinone the side chain is composed of a varying number of isoprenoid residues.

Physiology

Vitamin K is involved in the carboxylation of certain glutamate residues in proteins to form gamma-carboxyglutamate residues (abbreviated Gla-residues). The modified residues are often (but not always) situated within specific protein domains called Gla domains. Gla-residues are usually involved in binding calcium. The Gla-residues are essential for the biological activity of all known Gla-proteins.^[5]

At this time 14 human proteins with Gla domains have been discovered, and they play key roles in the regulation of three physiological processes:

- Blood coagulation: (prothrombin (factor II), factors VII, IX, X, protein C, protein S, and protein Z).^[6]
- Bone metabolism: osteocalcin, also called bone Gla-protein (BGP), and matrix gla protein (MGP).^[7]
- Vascular biology.^[8]

Like other liposoluble vitamins (A, D, E), vitamin K is stored in the fat tissue of the human body.

Recommended amounts

The U.S. Dietary Reference Intake (DRI) for an Adequate Intake (AI) of vitamin K for a 25-year old male is 120 micrograms/day. The Adequate Intake (AI) for adult women is 90 micrograms/day, for infants is 10–20 micrograms/day, for children and adolescents 15–100 micrograms/day. In 2002 it was found that to get maximum carboxylation of osteocalcin, one may have to take up to 1000 µg of vitamin K₁.^[9]

Toxicity

Although allergic reaction from supplementation is possible, there is no known toxicity associated with high doses of the phyloquinone (vitamin K₁) or menaquinone (vitamin K₂) forms of vitamin K and therefore no tolerable upper intake level (UL) has been set.

However, a synthetic form of vitamin K, vitamin K₃ (menadione), is demonstrably toxic. In fact, the FDA has banned this synthetic form of the vitamin from over-the-counter supplements because large doses have been shown to cause allergic reactions, hemolytic anemia, and cytotoxicity in liver cells.^[10]

Drug Interactions

Phylloquinone (K₁)^[11] ^[12] or menaquinone (K₂) are capable of blocking the blood thinning action of anticoagulants like warfarin, which work by interfering with the action of vitamin K. They also reverse the tendency of these drugs to cause arterial calcification in the long term.

Sources

Vitamin K₁ is found chiefly in leafy green vegetables such as spinach, swiss chard, and *Brassica* (*e.g.* cabbage, kale, cauliflower, broccoli, and brussels sprouts); some fruits such as avocado and kiwifruit are also high in vitamin K. By way of reference, two tablespoons of parsley contain 153% of the recommended daily amount of vitamin K.^[13] Some vegetable oils, notably soybean, contain vitamin K, but at levels that would require relatively large caloric consumption to meet the USDA recommended levels.^[14]

It is believed that phyloquinone's tight binding to the thylakoid membranes in the chloroplasts is the reason behind the poor bioavailability of vitamin K in green plants. For example, cooked spinach has a 5 percent bioavailability of phyloquinone. However when one adds fat to the spinach, the bioavailability increases to 13 percent due to the increased solubility of vitamin K in fat.^[15]

Menaquinone-4 and menaquinone-7 (vitamin K₂) are found in meat, eggs, dairy,^[16] and natto^[17]. MK-4 is synthesized by animal tissues, the rest (mainly MK-7) are synthesized by bacteria during fermentation. In natto 0%

of vitamin K is from MK-4 and in cheese 2–7%.^[18]

Deficiency

Average diets are usually not lacking in vitamin K and primary vitamin K deficiency is rare in healthy adults. As previously mentioned, newborn infants are at an increased risk of deficiency. Other populations with an increased prevalence of vitamin K deficiency include individuals who suffer from liver damage or disease (e.g. alcoholics), people with cystic fibrosis, inflammatory bowel diseases or those who have recently had abdominal surgeries. Groups that may suffer from secondary vitamin K deficiency include bulimics, those on stringent diets, and those taking anticoagulants. Other drugs that have been associated with vitamin K deficiency include salicylates, barbiturates, and cefamandole, although the mechanism is still unknown. There is no difference between the sexes as both males and females are affected equally. Symptoms of deficiency include heavy menstrual bleeding in women, anemia, bruising, and bleeding of the gums or nose.^[18]

Osteoporosis^{[19] [20]} and coronary heart disease^{[21] [22]} are strongly associated with lower levels of K₂ (menaquinone). Menaquinone is not inhibited by salicylates as happens with K₁, so menaquinone supplementation can alleviate the chronic vitamin K deficiency caused by long term aspirin use.

Biochemistry

Discovery

In 1929, Danish scientist Henrik Dam investigated the role of cholesterol by feeding chickens a cholesterol-depleted diet.^[23] After several weeks, the animals developed hemorrhages and started bleeding. These defects could not be restored by adding purified cholesterol to the diet. It appeared that—together with the cholesterol—a second compound had been extracted from the food, and this compound was called the coagulation vitamin. The new vitamin received the letter K because the initial discoveries were reported in a German journal, in which it was designated as *Koagulationsvitamin*. Edward Adelbert Doisy of Saint Louis University did much of the research that led to the discovery of the structure and chemical nature of vitamin K.^[24] Dam and Doisy shared the 1943 Nobel Prize for medicine for their work on vitamin K (K₁ and K₂) published in 1939. Several laboratories synthesized the compound(s) in 1939.^[25]

For several decades the vitamin K-deficient chick model was the only method of quantifying vitamin K in various foods: the chicks were made vitamin K-deficient and subsequently fed with known amounts of vitamin K-containing food. The extent to which blood coagulation was restored by the diet was taken as a measure for its vitamin K content. Three groups of physicians independently found this: Biochemical Institute, University of Copenhagen (Dam and Johannes Glavind), University of Iowa Department of Pathology (Emory Warner, Kenneth Brinkhous, and Harry Pratt Smith), and the Mayo Clinic (Hugh Butt, Albert Snell, and Arnold Osterberg).^[26]

The first published report of successful treatment with vitamin K of life-threatening hemorrhage in a jaundiced patient with prothrombin deficiency was made in 1938 by Smith, Warner, and Brinkhous.^[27]

Function

The function of vitamin K in the cell is to convert glutamate in proteins to gamma-carboxyglutamate (gla).

Within the cell, vitamin K undergoes electron reduction to a reduced form of vitamin K (called vitamin K hydroquinone) by the enzyme vitamin K epoxide reductase (or VKOR).^[28] Another enzyme then oxidizes vitamin K hydroquinone to allow carboxylation of Glu to Gla; this enzyme is called the gamma-glutamyl carboxylase^{[29] [30]} or the vitamin K-dependent carboxylase. The carboxylation reaction will only proceed if the carboxylase enzyme is able to oxidize vitamin K hydroquinone to vitamin K epoxide at the same time; the carboxylation and epoxidation reactions are said to be coupled reactions. Vitamin K epoxide is then re-converted to vitamin K by vitamin K

epoxide reductase. These two enzymes comprise the so-called vitamin K cycle.^[31] One of the reasons humans are rarely deficient in vitamin K is that vitamin K is continually recycled in our cells.

Warfarin and other coumarin drugs block the action of the vitamin K epoxide reductase.^[32] This results in decreased concentrations of vitamin K and vitamin K hydroquinone in the tissues, such that the carboxylation reaction catalyzed by the glutamyl carboxylase is inefficient. This results in the production of clotting factors with inadequate Gla. Without Gla on the amino termini of these factors, they no longer bind stably to the blood vessel endothelium and cannot activate clotting to allow formation of a clot during tissue injury. As it is impossible to predict what dose of warfarin will give the desired degree of suppression of the clotting, warfarin treatment must be carefully monitored to avoid over-dosing. (See the warfarin article.)

Gamma-carboxyglutamate proteins, or Gla-proteins

At present, the following human Gla-containing proteins have been characterized to the level of primary structure: the blood coagulation factors II (prothrombin), VII, IX, and X, the anticoagulant proteins C and S, and the Factor X-targeting protein Z. The bone Gla-protein osteocalcin, the calcification inhibiting matrix gla protein (MGP), the cell growth regulating growth arrest specific gene 6 protein (Gas6), and the four transmembrane Gla proteins (TMGPs) the function of which is at present unknown. Gas6 can function as a growth factor that activates the Axl receptor tyrosine kinase and stimulates cell proliferation or prevents apoptosis in some cells. In all cases in which their function was known, the presence of the Gla-residues in these proteins turned out to be essential for functional activity.

Gla-proteins are known to occur in a wide variety of vertebrates: mammals, birds, reptiles, and fish. The venom of a number of Australian snakes acts by activating the human blood clotting system. Remarkably, in some cases activation is accomplished by snake Gla-containing enzymes that bind to the endothelium of human blood vessels and catalyze the conversion of procoagulant clotting factors into activated ones, leading to unwanted and potentially deadly clotting.

Another interesting class of invertebrate Gla-containing proteins is synthesized by the fish-hunting snail *Conus geographus*.^[33] These snails produce a venom containing hundreds of neuro-active peptides, or conotoxins, which is sufficiently toxic to kill an adult human. Several of the conotoxins contain 2–5 Gla residues.^[34]

Methods of assessment

Prothrombin time test:

- Measures the time required for blood to clot
- Blood sample mixed with citric acid and put in a fibrometer.
- Delayed clot formation indicates a deficiency.

Unfortunately insensitive to mild deficiency as the values do not change until the concentration of prothrombin in the blood has declined by at least 50 percent^[35]

Plasma Phylloquinone:

- Was found to be positively correlated with phylloquinone intake in elderly British women, but not men^[36]

However an article by Schurges et al. reported no correlation between FFQ and plasma phylloquinone^[37]

Urinary γ -carboxyglutamic acid:

- Urinary Gla responds to changes in dietary vitamin K intake.
- Several days are required before any change can be observed.

In a study by Booth et al. increases of phylloquinone intakes from 100 μg to between 377–417 μg for 5 days did *not* induce a significant change Response may be age-specific^[38]

Function in bacteria

Many bacteria, such as *Escherichia coli* found in the large intestine, can synthesize vitamin K₂ (menaquinone),^[39] but not vitamin K₁ (phylloquinone). In these bacteria, menaquinone will transfer two electrons between two different small molecules, in a process called anaerobic respiration.^[40] For example, a small molecule with an excess of electrons (also called an electron donor) such as lactate, formate, or NADH, with the help of an enzyme, will pass two electrons to a menaquinone. The menaquinone, with the help of another enzyme, will in turn transfer these 2 electrons to a suitable oxidant, such fumarate or nitrate (also called an electron acceptor). Adding two electrons to fumarate or nitrate will convert the molecule to succinate or nitrite + water, respectively. Some of these reactions generate a cellular energy source, ATP, in a manner similar to eukaryotic cell aerobic respiration, except that the final electron acceptor is not molecular oxygen, but say fumarate or nitrate (In aerobic respiration, the final oxidant is molecular oxygen (O₂), which accepts four electrons from an electron donor such as NADH to be converted to water.) *Escherichia coli* can carry out aerobic respiration and menaquinone-mediated anaerobic respiration.

Vitamin K injection in newborns

The blood clotting factors of newborn babies are roughly 30 to 60 percent that of adult values; this may be due to the reduced synthesis of precursor proteins and the sterility of their guts. Human milk contains between 1 and 4 micrograms/litre of vitamin K₁, while formula derived milk can contain up to 100 micrograms/litre in supplemented formulas. Vitamin K₂ concentrations in human milk appear to be much lower than those of vitamin K₁. It is estimated that there is a 0.25 to 1.7 percent occurrence of vitamin K deficiency bleeding in the first week of the infant's life with a prevalence of 2-10 cases per 100,000 births.^[41] Premature babies have even lower levels of the vitamin and are at a higher risk from this deficiency.

USA

As a result of the occurrences of vitamin K deficiency bleeding, the Committee on Nutrition of the American Academy of Pediatrics has recommended that 0.5 to 1.0 mg vitamin K₁ be administered to all newborns shortly after birth.^[42]

UK

In the UK, vitamin K is administered to newborns as either a single injection at birth or three orally administered doses given at birth and then over the baby's first month.

Controversy

Controversy arose in the early 1990s regarding this practice when two studies were shown suggesting a relationship between parenteral administration of vitamin K and childhood cancer (14). However, poor methods and small sample sizes led to the discrediting of these studies and a review of the evidence published in 2000 by Ross and Davies found no link between the two.^[43]

Dietary Vitamin K and bone health

There is physiological and observational evidence that Vitamin K plays a role in bone growth and the maintenance of bone density, but efforts to delay the onset of osteoporosis by Vitamin K supplementation have proven ineffective. Biophysical studies have indicated that supplemental vitamin K promotes osteotrophic processes and slows osteoclastic processes via calcium bonding. Other studies have shown that vitamin K antagonists (usually a class of anticoagulants) lead to early calcification of the epiphysis and epiphysial line in mice and other animals, causing seriously decreased bone growth. This is due to defects in osteocalcin and matrix gla protein. Their primary function is to prevent overcalcification of the bone and cartilage. Vitamin K is important in the process of carboxylating

glutamic acid (Glu) in these proteins to gamma-carboxyglutamic acid (Gla), which is necessary for their function.^[44]
^[45]

Data from the 1998 Nurses Health Study, an observational study, indicated an inverse relationship between dietary vitamin K₁ and the risk of hip fracture. After being given 110 micrograms/day of vitamin K, women who consumed lettuce one or more times per day had a significantly lower risk of hip fracture than women who consumed lettuce one or fewer times per week. In addition to this, high intakes of vitamin D but low intakes of vitamin K were suggested to pose an increased risk of hip fracture.^[46] ^[45]

However, a large multicentre randomized placebo-controlled trial was performed to test the supplementation of Vitamin K in post-menopausal women with osteopenia. Despite heavy doses of Vitamin K, no differences were found in bone density between the supplemented and placebo groups.^[47]

In Japan, a form of vitamin K₂ is recognized as a treatment for osteoporosis.^[48] ^[46] However the long term effects and benefits are unknown and it remains controversial.

Vitamin K and Alzheimer's disease

Research into the antioxidant properties of vitamin K indicates that the concentration of vitamin K is lower in the circulation of carriers of the APOE4 gene, and recent studies have shown its ability to inhibit nerve cell death due to oxidative stress. It has been hypothesized that vitamin K may reduce neuronal damage and that supplementation may hold benefits to treating Alzheimer's disease, although more research is necessary in this area.^[49]

Vitamin K used topically

Vitamin K may be applied topically, typically as a 5% cream, to diminish postoperative bruising from cosmetic surgery and injections, broken capillaries (spider veins), to treat rosacea and to aid in the fading of hyperpigmentation and dark under-eye circles.

Vitamin K and cancer

While researchers in Japan were studying the role of vitamin K₂ in the prevention of bone loss in females with liver disease, they discovered another possible effect of this phytonutrient. This two year study which involved 21 women with viral liver cirrhosis found that women in the supplement group were 90 percent less likely to develop liver cancer.^[50] ^[51] A German study performed on men with prostate cancer found a significant inverse relationship between vitamin K₂ consumption and advanced prostate cancer.^[52]

Vitamin K as antidote for poisoning by 4-hydroxycoumarin drugs

Vitamin K is a true antidote for poisoning by 4-hydroxycoumarin anticoagulant drugs (sometimes loosely referred to as coumarins). These include the pharmaceutical warfarin, and also anticoagulant-mechanism poisons such as bromadiolone, which are commonly found in rodenticides. 4-hydroxycoumarin drugs possess anticoagulatory and rodenticidal properties because they can completely block synthesis of vitamin K in the liver. Death is usually a result of internal hemorrhage. Treatment usually consists of repeated intravenous doses of vitamin K, followed by doses in pill form for a period of at least two weeks, though possibly up to 2 months, afterwards (in the case of the more potent 4-hydroxycoumarins used as rodenticides). If caught early, prognosis is good, even when great amounts of the drug or poison are ingested.

History of discovery

The precise function of vitamin K was not discovered until 1974, when three laboratories (Stenflo *et al.*^[53], Nelsestuen *et al.*^[54], and Magnusson *et al.*^[55]) isolated the vitamin K-dependent coagulation factor prothrombin (Factor II) from cows that received a high dose of a vitamin K antagonist, warfarin. It was shown that while warfarin-treated cows had a form of prothrombin that contained 10 glutamate amino acid residues near the amino terminus of this protein, the normal (untreated) cows contained 10 unusual residues that were chemically identified as gamma-carboxyglutamate, or Gla. The extra carboxyl group in Gla made clear that vitamin K plays a role in a carboxylation reaction during which Glu is converted into Gla.

The biochemistry of how vitamin K is used to convert Glu to Gla has been elucidated over the past thirty years in academic laboratories throughout the world.

External links

- Jane Higdon, "Vitamin K^[56]", Micronutrient Information Center, *Linus Pauling Institute*
- Vitamin K: Another Reason to Eat Your Greens^[57]
- Vitamin K: Signs of Deficiency^[58]
- Vitamin K: Vitamin Deficiency, Dependency, and Toxicity^[59]: Merck Manual Professional
- An Alternative Perspective on Vitamin K Prophylaxis^[60]
- Health Benefits of Vitamin K₂^[61]
- Vitamin K content^[62]: USDA National Nutrient Database for Standard Reference, Release 19

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